Jochen Ellermann,*a Adolf Veit,a Ekkehard Lindner,*b and Sigurd Hoehneb

 Institut f
ür Anorganische Chemie, Universit
ät Erlangen-N
ürnberg, Egerlandstrasse 1, D-8520 Erlangen, West Germany

Institut für Anorganische Chemie, Universität Tübingen, Auf der Morgenstelle 18, D-7400 Tübingen, West Germany

Reaction of $N[(CH_2)_2As(C_6H_5)_2]_3$ (1) with gaseous HI leads to $\{HN[(CH_2)_2Asl_2]_3\}I$ (2) and hydrolysis of (2) with H_2O-NH_3 in tetrahydrofuran gives $[N(CH_2CH_2)_3]_8(As_4O_4)_6$ (3); single crystal X-ray diffraction shows that one molecule of (3) contains six eight-membered As_4O_4 rings connected together by triethylamine units.

In several papers we have reported on organoarsenic cage compounds with nortricyclene, noradamantane, and adamantane structures.¹⁻⁶ The present paper describes the synthesis and the crystal and molecular structure of a macrocyclic arsa-aza-oxa-cryptand which, as a special feature, contains six crown-shaped, waved, eight-membered As_4O_4 rings (similar to sulphur- S_8 rings).

When $\{HN[(CH_2)_2AsI_2]_3\}I(2)$, which is accessible, according to equation (1), by selective cleavage of phenyl groups⁸ from $N[(CH_2)_2As(C_6H_5)_2]_3(1)^{7,9}$ with HI, was allowed to react with H_2O-NH_3 in tetrahydrofuran (THF), according to equation (2) the macrocyclic arsa-aza-oxa-cryptand (3) was formed. After addition of conc. NH_3 to a suspension of (2) in

$$N[(CH_{2})_{2}As(C_{6}H_{5})_{2}]_{3} + 7HI \xrightarrow{CH_{2}Cl_{2}} \{HN[(CH_{2})_{2}AsI_{2}]_{3}\}I + (1)$$
(2)

6C H (1)

 $8(2) + 24H_2O + 56NH_3 \xrightarrow{\text{THF}} [N(CH_2CH_2)_3]_8(As_4O_4)_6 + (3)$ $56NH_4I \quad (2)$



Figure 1. General view of a molecule of (3) with the atoms represented by their 50% probability ellipsoids for thermal motion.

THF an immediate decolourization of the yellow reaction mixture occured to give a clear solution, from which (3) precipitated after evaporation of THF at room temperature [63.5%, colourless crystals, $v(As_4O_4)$ 740vs and v(AsC)596s cm⁻¹]. In a mass spectrometer (3) could not be evaporated without decomposition, however, the molecular mass was established by osmometric determination. The ¹H n.m.r. spectrum of (3) ([²H₅]pyridine, SiMe₄ int.) reveals two multiplets at δ 2.35 (72H, N-CH[H] and As-CH₂) and 3.20 (24H, N-CH[H]). The down-field shift of the 24 protons of the *N*-bonded methyl groups can be explained from the organo-



Figure 2. Projection from above of an arsenic–oxygen eightmembered ring. Distances (pm): As(1)–O(1) 181(2), O(1)–As(2) 173(2), As(2)–O(2) 175(3), O(2)–As(3) 183(2), As(3)–O(3) 176(3), O(3)–As(4) 181(3), As(4)–O(4) 181(2), O(4)–As(1) 183(2), As(1)–C(11) 205(4), As(2)–C(21) 192(3), As(3)–C(31) 205(4), As(4)–C(41) 177(4), C(11)–C(12) 152(6), C(21)–C(22) 148(6), C(31)–C(32) 156(5), C(41)–C(42) 158(8), C(12)–N(3') 150(6), C(22)–N(6') 141(6), C(32)–N(3) 153(4), and C(42)–N(4) 159(5). Angles (°): O(4)–As(1)–O(1) 102(1), O(4)–As(1)–C(11) 92(1), O(1)–As(1)–C(11) 96(1), O(1)–As(2)–O(2) 102(1), O(1)–As(2)–C(21) 95(1), O(2)–As(3)–O(3) 101(1), O(2)–As(3)–C(31) 93(1), O(3)–As(3)–C(31) 96(2), O(3)–As(4)–C(41) 94(1), As(1)–O(1)–As(2) 119(1), As(2)–O(2)–As(3) 118(1), As(3)–O(3)–As(4) 118(1), As(4)–O(4)–As(1) 115(1), As(1)–C(11) 102(2), As(4)–C(41) 94(2), C(41)–C(22) 113(4), C(11)–C(12)–N(3') 115(3), C(21)–C(22)–N(6') 110(3), C(31)–C(32)–N(3) 112(3), and C(41)–C(42)–N(4) 108(3).

arsenic cage structure of (3) by the interaction of these protons with the oxygen atoms of the As_4O_4 rings.

Crystal data: $C_{48}H_{96}As_{24}N_8O_{24}$, M = 2967.46, hexagonal, space group $P\overline{3}$, a = 1565.5(5), c = 2336.2(8) pm, Z = 2, $D_c = 1.987$ g cm⁻³, $D_m = 1.995$ g cm⁻³, F(000) = 2848, $\mu(Mo-K_{\alpha}) = 78.1$ cm⁻¹. The structure was refined to R = 0.087 for 1831 independent reflections $[I \ge 3\sigma(I), 6 \le 2\theta \le 50^{\circ},$ Enraf-Nonius CAD-4].†

The X-ray structure shows that one molecule of (3) consists of six eight-membered As_4O_4 rings which are connected together by the triethylamine units linked to arsenic. The eight nitrogen atoms occupy the corners of a trigonally distorted cube, on the surfaces of which the crown-shaped arsenicoxygen rings are roughly located (Figure 1). One of these rings is shown in Figure 2 from above and for illustration of the ring conformation a side-view (Figure 3) is also given.

The atoms N(4) and N(7) are located on a crystallographic threefold axis, so that the molecule has C_3 symmetry. In comparison with the sum of the covalent radii¹⁰ and other similar examples,¹¹ the As–O bonds having an average length of



Figure 3. Projection of a side view of an arsenic-oxygen eightmembered ring.

[†] The structure was solved by direct methods and Fourierdifference synthesis, and anisotropically refined using the SHELX-76 program [except for H, C(81), and C(82)]. The atomic co-ordinates for this work are available on request from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cambridge CB2 1EW. Any request should be accompanied by the full literature citation for this communication. Further evidence for the presence of As-O bonds with partial π character is given by the enlarged As-O-As and O-As-O angles (115—120° and 99—104°, respectively), whereas those angles which involve carbon atoms are found, as expected, between 92 and 97°.

We gratefully acknowledge the generous support of this research work by the Deutsche Forschungsgemeinschaft, the Fonds der Chemischen Industrie, and Hoechst AG, Frankfurt/Main.

Received, 29th December 1981, Com. 1478

References

- 1 J. Ellermann and H. Schössner, Angew. Chem., 1974, 86, 646; Angew. Chem., Int. Ed. Engl., 1974, 13, 601.
- 2 G. Thiele, G. Zoubek, H. A. Lindner, and J. Ellermann, Angew. Chem., 1978, 90, 133; Angew. Chem., Int. Ed. Engl., 1978, 17, 135.
- 3 J. Ellermann, M. Lietz, P. Merbach, G. Thiele, and G. Zoubek, Z. Naturforsch., Teil B, 1979, 34, 975.
- 4 J. Ellermann and M. Lietz, Z. Naturforsch., Teil B, 1980, 35, 1514.
- 5 J. Ellermann and M. Lietz, J. Organomet. Chem., 1981, 215, 165.
- 6 J. Ellermann and M. Lietz, Z. Naturforsch., Teil B, 1981, 36, 1532.
- 7 J. Ellermann and A. Veit, unpublished results.
- 8 J. Ellermann, H. Schössner, A. Haag, and H. Schödel, J. Organomet. Chem., 1974, 65, 33.
- 9 L. Sacconi, I. Bertini, and F. Mani, *Inorg. Chem.*, 1968, 7, 1417.
- 10 L. Pauling, 'Die Natur der Chemischen Bindung,' Edn. 3, Verlag Chemie, Weinheim/Bergstr., 1968.
- 11 J. v. Seyerl, B. Sigwarth, and G. Huttner, Chem. Ber., 1981, 114, 1407.
- 12 M. Jakob and E. Weiss, J. Organomet. Chem., 1978, 153, 31.
- 13 R. Battaglia, H. Kirsch, C. Krüger, and L.-K. Liu, Z. Naturforsch., Teil B, 1980, 35, 719.